

DOWEX[™] MONOSPHERE[™] 88 (H)

Ion Exchange Resin for Sweetener Applications

Product	Туре	Matrix	Functional group
DOWEX™ MONOSPHERE™ 88 (H)	Strong acid cation	Styrene-DVB, macroporous	Sulfonate

lonic form as produced		H⁺
Total exchange capacity, min.	eq/L	1.7
Water content	%	50 - 56
Bead size distribution Volume median diameter 400 - 720 μm, min.	μm %	500 - 600 95
Total swelling (Na ⁺ \rightarrow H ⁺)	%	5
Whole uncracked beads, min.	%	95
Particle density	g/mL	1.2
Shipping weight**	g/L Ibs/ft ³	770 48

aximum operating temperature (H⁺ form)	93°C (200°F)
I range	0 - 14
ed depth, min.	91 cm (3 ft)
ow rates: Service Backwash Regeneration time, minimum Displacement rinse, minimum	2 - 4 bed volumes/hour See Figure 1 30 - 45 min. 30 - 45 min.
tal rinse requirement (new)	2 - 5 bed volumes
egenerant: Concentration evel, 100% basis [†] Temperature, max.	7% HCl 5 - 6 lbs/ft ³ 80 - 96 kg/m ³ 93°C (200°F)
	I range ed depth, min. pow rates: ervice ackwash Regeneration time, minimum bisplacement rinse, minimum tal rinse requirement (new) egenerant: concentration evel, 100% basis [†]

[†] Regeneration level may be lower for counter-current regeneration systems.

** As per the backwashed and settled density of the resin, determined by ASTM D-2187.

Recommended

Operating Conditions

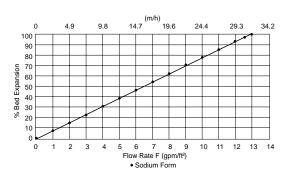
Typical Properties and Applications

DOWEX[™] MONOSPHERE[™] 88 (H) resin is a strong acid cation resin made using a Dowpatented process which produces beads with remarkable size uniformity. Chemically optimized for syrup processing, the resin provides an ideal balance of high operating capacity, excellent physical strength, economical regeneration and long resin life, which together, result in low operating costs.

Packaging

5 cubic feet fiber drums or 1 cubic meter super sacks

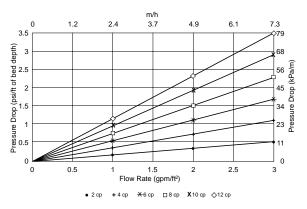
Figure 1. Backwash Expansion Data (Exhausted)



For other temperatures use:

$$\begin{split} F_T &= F_{77^\circ F} \; [1+\; 0.008 \; (T_{^\circ F} \; -77)], \; \text{where } F \equiv gpm/ft^2 \\ F_T &= F_{25^\circ C} \; [1+\; 0.008 \; (1.8T_{^\circ C} \; -45)], \; \text{where } F \equiv m/h \end{split}$$

Figure 2. Pressure Drop Data



For other temperatures use:

 $P_T = P_{20^{\circ}C} / (0.026 T_{\circ C} + 0.48)$, where P = bar/m $P_T = P_{68^{\circ}F} / (0.014 T_{\circ F} + 0.05)$, where P = psi/ft

DOWEX™ Ion Exchange ResinsFor more information about DOWEXresins, call the Dow Water Solutionsbusiness:1-800-447-4369Latin America:(+55) 11-5188-9222Europe:(+32) 3-450-2240Pacific:+60 3 7958 3392Japan:+813 5460 2100

http://www.dowwatersolutions.com

+86 21 2301 9000

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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